

The role of the argon concentration on the evolution of morphology and structural properties of boron doped diamond

Bárbara da Silva Pinheiro¹, Luiz Carlos Rosa², Denis Angelo da Silva³, Jossano Saldanha Marcuzzo^{1,4}, Emerson Sarmiento Gonçalves⁵, Mauricio Ribeiro Baldan²

¹Instituto Nacional de pesquisas espaciais, ²Instituto Nacional de Pesquisas Espaciais, ³Universidade Federal de São Paulo, ⁴Faculdade de Tecnologia de São Jose dos Campos, ⁵Instituto de Aeronáutica e Espaço

e-mail: barbarapinheirosjc@outlook.com

Boron-doped diamond films have been grown by hot filament chemical vapor deposition reactor in Ar/H₂/CH₄ gas mixtures. Diamond films were grown using different argon concentrations X=0,10,30,40,60, with (99-X)% H₂ and 1%CH₄. An additional H₂ line passing through a bubbler containing B₂O₃ dissolved in metanol was used to carry the boron during the doping process. The substrate temperature was kept at 630 °C. The electrically heated tungsten filament positioned under the substrate surface was used to activate the gas mixture. The pressure and flow of the gas mixture Ar/H₂/CH₄ were kept constant during all the deposition experiments. The deposition time was about 16 h. Three film sample sets were produced a) without dopping; b) without doping with the additional H₂ line passing through metanol solution and c) with doping. This investigation describes separately the influence of argon, metanol solution and boron solution on the morphology and structural properties of the diamond films. Raman Scattering Spectroscopy confirmed growth of diamond with the characteristic peak located at 1332 cm⁻¹ and the doping of film with the appearance of bands at 1220 and 500 cm⁻¹, as well as the presence of graphite due to G-band located at 1580 cm⁻¹. The micrographs of scanning electron microscopy reveal that the surface morphology, the grain size, and the growth mechanism of the diamond films depend strongly on the ratio of Ar to H₂ in the reactant gases. The addition of argon to the growth gas mixture clearly revealed the transition from microcrystalline to nanocrystalline diamond films. The additional H₂ line through the metanol solution (without B₂O₃) did not provides a substantial changes in the morphology and structural properties. As far as boron doped diamond films concern, from the separation of the three contributions it was possible to attribute the existence of additional bands in the Raman spectra only due to the presence of B₂O₃.